The author's approach is effective for mastering the details of thermodynamic calculations. This book is particularly recommended for studying thermodynamics without the benefit of an instructor.

A. L. MYERS
UNIVERSITY OF PENNSYLVANIA

ERRATA

In "A Theoretical Approach to Nonfoaming Adsorptive Bubble Fractionation" by Robert Lemlich [Vol. 12, No. 4, pp. 802-804 (1966)], in the first paragraph under the heading Theory, the words *rich* and *lean* should be interchanged.

In "Hot-Film Anemometry Measurements of Turbulence in Pipe Flow: Organic Solvents," by G. K. Patterson and J. L. Zakin [Vol. 13, No. 3, pp. 513-519 (1967)], the limits of the integral in Equation (2) should be changed from 0 to r, to r to a.

(Continued from page 834)

In the discussion of chemical reactions in Chapter 4, the author develops the concept of spontaneous ($\Delta G^o < 0$) and forbidden ($\Delta G^o > 0$) reactions. The meaning is that total conversion is either spontaneous or forbidden. Unfortunately it is not emphasized in Chapter 4 that partial conversion is spontaneous regardless of the sign of ΔG^o. Therefore the reader is left with the impression that chemical reactions do not proceed if $\Delta G^o > 0$.

There are less than one dozen typographical errors in this book, quite an achievement considering the large number of detailed examples.